## **Amendments to the Claims**

This listing of claims will replace all prior versions, and listings, of claims in the application.

## **Listing of Claims**

Claim 1. (Original) A process for partially oxidizing propene to acrylic acid in the gas phase under heterogeneous catalysis by conducting a starting reaction gas mixture 1 which comprises propene, molecular oxygen and at least one inert gas, and contains the molecular oxygen and the propene in a molar  $O_2:C_3H_6$  ratio of  $\geq 1$  in a first reaction stage over a fixed catalyst bed 1 which is arranged in two spatially successive reaction zones A, B, the temperature of reaction zone A being a temperature in the range from 290 to 380° C and the temperature of reaction zone B likewise being a temperature in the range from 290 to 380° C, and whose active composition is at least one multimetal oxide comprising the elements Mo, Fe and Bi, in such a way that reaction zone A extends to a propene conversion of from 40 to 80 mol % and the propene conversion on single pass through the fixed catalyst bed 1 is ≥ 90 mol % and the accompanying selectivity of acrolein formation and also of acrylic acid by-production taken together is ≥ 90 mol %, the temperature of the product gas mixture leaving the first reaction stage is optionally reduced by cooling and molecular oxygen and/or inert gas are optionally added to the product gas mixture, and then the product gas mixture, as a starting reaction gas mixture 2 comprising acrolein, molecular oxygen and at least one inert gas and containing the molecular oxygen and the acrolein in a molar  $O_2:C_3H_4O$  ratio of  $\geq 0.5$ , is conducted in a second reaction stage over a fixed catalyst bed 2 which is arranged in two spatially successive reaction zones C, D, the temperature of reaction zone C being a temperature in the range from 230 to 320° C and the temperature of reaction zone D likewise being a temperature in the range from 230 to 320° C, and whose active composition is at least one multimetal oxide comprising the elements Mo and V, in such a way that reaction zone C extends to an acrolein conversion of from 45 to 85 mol % and the acrolein conversion on single pass through the fixed catalyst bed 2 is  $\geq$  90 mol % and the selectivity of acrylic acid formation assessed over all reaction zones, based on converted propene, is  $\geq$  80 mol %, the sequence in time in which the reaction gas mixture flows through the reaction zones corresponding to the alphabetic sequence of the reaction zones, wherein

- a) the hourly space velocity of the propene contained in the starting reaction gas mixture 1 on the fixed catalyst bed 1 is < 160 l (STP) of propene/l of fixed catalyst bed 1  $\circ$  h and  $\geq$  90 l (STP) of propene/l of fixed catalyst bed 1  $\circ$  h;
- b) the volume-specific activity of the fixed catalyst bed 1 is either constant or increases at least once in the flow direction of the reaction gas mixture over the fixed bed catalyst bed 1;
- c) the difference  $T^{maxA} T^{maxB}$ , formed from the highest temperature  $T^{maxA}$  which the reaction gas mixture has within reaction zone A and the highest temperature  $T^{maxB}$  which the reaction gas mixture has within reaction zone B is  $\geq 0^{\circ}$  C;
- d) the hourly space velocity of the acrolein contained in the starting reaction gas mixture 2 on the fixed catalyst bed 2 is < 145 l (STP) of acrolein/l of fixed catalyst bed 2  $\circ$  h and  $\geq$  70 l (STP) of acrolein/l of fixed catalyst bed 2  $\circ$  h;

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- e) the volume-specific activity of the fixed catalyst bed 2 increases at least once in the flow direction of the reaction gas mixture over the fixed bed catalyst bed 2; and
- f) the difference  $T^{maxC} T^{maxD}$ , formed from the highest temperature  $T^{maxC}$  which the reaction gas mixture has within reaction zone C and the highest temperature  $T^{maxD}$  which the reaction gas mixture has within reaction zone D, is  $\geq 0^{\circ}$  C.
- Claim 2. (Original) A process as claimed in claim 1, wherein the difference  $T^{maxA}$   $T^{maxB}$  is  $\geq 3^{\circ}$  C and  $\leq 70^{\circ}$  C.
- Claim 3. (Original) A process as claimed in claim 1, wherein the difference  $T^{maxA}$   $T^{maxB}$  is  $\geq 20^{\circ}$  C and  $\leq 60^{\circ}$  C.
- Claim 4. (Currently Amended) A process as claimed in any of claims claim1 to 3, wherein the difference  $T^{maxC} T^{maxD}$  is  $\geq 3^{\circ}$  C and  $\leq 60^{\circ}$  C.
- Claim 5. (Currently Amended) A process as claimed in any of claims claim1 to 3, wherein the difference  $T^{maxC} T^{maxD}$  is  $\geq 3^{\circ}$  C and  $\leq 40^{\circ}$  C.
- Claim 6. (Currently Amended) A process as claimed in any of claims claim 1 to 5, wherein the hourly space velocity of the propene contained in the starting gas mixture on the fixed catalyst bed 1 is  $\geq$ 1001 (STP) or propene/l h and  $\leq$  150 1 (STP) or propene/l h.

Claim 7. (Currently Amended) A process as claimed in any-of-claims claim 1 to 5, wherein the hourly space velocity of the propene contained in the starting gas mixture on the fixed catalyst bed 1 is  $\geq$ 1101 (STP) or propene/l • h and  $\leq$  145 1 (STP) or propene/l •h.

Claim 8. (Currently Amended) A process as claimed in any of claims claim 1 to 7, wherein the difference  $T_B - T_A$  between the temperature of reaction zone B,  $T_B$ , and the temperature of reaction zone A,  $T_A$ , is  $\geq -10^{\circ}$  C and  $\leq 0^{\circ}$  C.

Claim 9. (Currently Amended) A process as claimed in any of claims claim 1 to 8, wherein the difference  $T_C - T_D$  between the temperature of reaction zone D,  $T_D$ , and the temperature of reaction zone C,  $T_C$ , is  $\geq -10^{\circ}$  C and  $\leq 0^{\circ}$  C.

Claim 10. (Currently Amended) A process as claimed in any of claims claim 1 to 9, wherein the temperature of reaction zone A is from 305 to 365° C.

Claim 11. (Currently Amended) A process as claimed in any of claims claim 1 to 9, wherein the temperature of reaction zone A is from 310 to 340° C.

Claim 12. (Currently Amended) A process as claimed in any of claims claim 1 to 11, wherein the temperature of reaction zone C is from 250 to 300° C.

Claim 13. (Currently Amended) A process as claimed in any of-claims claim 1 to 12, wherein the temperature of reaction zone C is from 260 to 280° C.

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Claim 14. (Currently Amended) A process as claimed in any of claims claim 1 to 13, wherein the active composition of the fixed catalyst bed 1 is at least one multimetal oxide active composition of the general formula I

$$Mo_{12}Bi_aFe_bX^1_cX^2_dX^3_eX^4_fO_n$$
 (I)

where

 $X^1$  = nickel and/or cobalt,

 $X^2$  = thallium, an alkali metal and/or an alkaline earth metal,

X<sup>3</sup> = zinc, phosphorus, arsenic, boron, antimony, tin, cerium, lead and/or tungsten,

X<sup>4</sup> = silicon, aluminum, titanium and/or zirconium,

a = from 0.5 to 5,

b = from 0.01 to 5,

c = from 0 to 10,

d = from 0 to 2,

e = from 0 to 8,

f = from 0 to 10 and

n = a number which is determined by the valency and frequency of the elements other than oxygen in I.

Claim 15. (Currently Amended) A process as claimed in any of claims claim 1 to 14, wherein the active composition of the fixed catalyst bed 2 is at least one multimetal oxide active composition of the general formula IV

$$Mo_{12}V_aX_b^1X_c^2X_d^3X_e^4X_f^5X_g^6O_n$$
 (IV)

where the variables are defined as follows:

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X^1 = W, Nb, Ta, Cr and/or Ce,
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$$X^2$$
 = Cu, Ni, Co, Fe, Mn and/or Zn,

$$X^3$$
 = Sb and/or Bi,

$$X^4$$
 = one or more alkali metals,

$$X^5$$
 = one or more alkaline earth metals,

$$X^6$$
 = Si, Al, Ti and/or Zr,

$$a = from 1 to 6,$$

b = from 
$$0.2$$
 to  $4$ ,

$$c = from 0.5 to 18,$$

$$d = from 0 to 40,$$

$$e = from 0 to 2,$$

$$f = from 0 to 4,$$

$$g = from 0 to 40 and$$

n = a number which is determined by the valency and frequency of the elements other than oxygen in IV.